



## Effects of CO<sub>2</sub> perturbation on phosphorus pool sizes and uptake in a mesocosm experiment during a low productive summer season in the northern Baltic Sea

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**Abstract.** Studies investigating the effect of increasing CO<sub>2</sub> levels on the phosphorus cycle in natural waters are lacking although phosphorus often controls phytoplankton development in many aquatic systems. The aim of our study was to analyse effects of elevated CO<sub>2</sub> levels on phosphorus pool sizes and uptake. The phosphorus dynamic was followed in a CO<sub>2</sub>-manipulation mesocosm experiment in the Storfjärden (western Gulf of Finland, Baltic Sea) in summer 2012 and was also studied in the surrounding fjord water. In all mesocosms as well as in surface waters of Storfjärden, dissolved organic phosphorus (DOP) concentrations of  $0.26 \pm 0.03$  and  $0.23 \pm 0.04 \mu\text{mol L}^{-1}$ , respectively, formed the main fraction of the total P-pool (TP), whereas phosphate (PO<sub>4</sub>) constituted the lowest fraction with mean concentration of  $0.15 \pm 0.02$  in the mesocosms and  $0.17 \pm 0.07 \mu\text{mol L}^{-1}$  in the fjord. Transformation of PO<sub>4</sub> into DOP appeared to be the main pathway of PO<sub>4</sub> turnover. About 82 % of PO<sub>4</sub> was converted into DOP whereby only 18 % of PO<sub>4</sub> was transformed into particulate phosphorus (PP). PO<sub>4</sub> uptake rates measured in the mesocosms ranged between 0.6 and  $3.9 \text{ nmol L}^{-1} \text{ h}^{-1}$ . About 86 % of them was

realized by the size fraction  $< 3 \mu\text{m}$ . Adenosine triphosphate (ATP) uptake revealed that additional P was supplied from organic compounds accounting for 25–27 % of P provided by PO<sub>4</sub> only. CO<sub>2</sub> additions did not cause significant changes in phosphorus (P) pool sizes, DOP composition, and uptake of PO<sub>4</sub> and ATP when the whole study period was taken into account. However, significant short-term effects were observed for PO<sub>4</sub> and PP pool sizes in CO<sub>2</sub> treatments  $> 1000 \mu\text{atm}$  during periods when phytoplankton biomass increased. In addition, we found significant relationships (e.g., between PP and Chl *a*) in the untreated mesocosms which were not observed under high *f*CO<sub>2</sub> conditions. Consequently, it can be hypothesized that the relationship between PP formation and phytoplankton growth changed with CO<sub>2</sub> elevation. It can be deduced from the results, that visible effects of CO<sub>2</sub> on P pools are coupled to phytoplankton growth when the transformation of PO<sub>4</sub> into POP was stimulated. The transformation of PO<sub>4</sub> into DOP on the other hand does not seem to be affected. Additionally, there were some indications that cellular mechanisms of P regulation might be modified un-

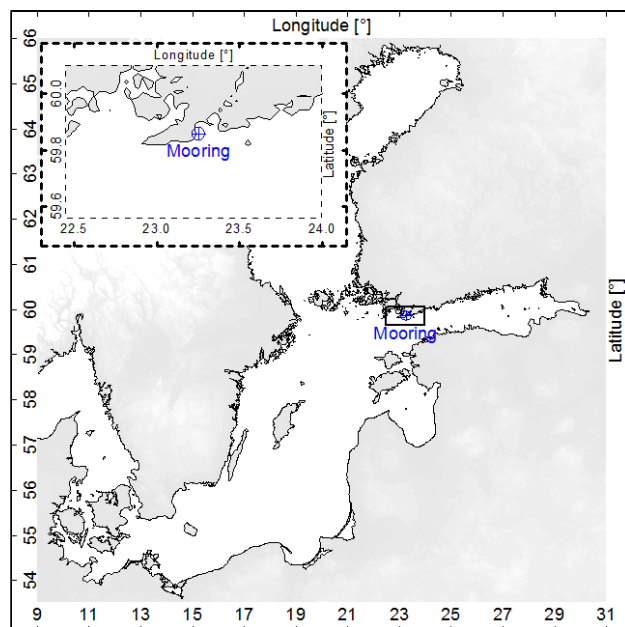
der CO<sub>2</sub> elevation changing the relationship between cellular constituents.

## 1 Introduction

Increasing emissions of anthropogenic CO<sub>2</sub> into the atmosphere and subsequent acidification of the ocean can potentially affect the diversity of organisms and the functioning of marine ecosystems (Eisler, 2012). The rise in atmospheric CO<sub>2</sub> concentrations was accelerated from  $3.4 \pm 0.2$  in the 1980s to  $4.0 \pm 0.2$  PgC yr<sup>-1</sup> in the 2000s leading to increases of CO<sub>2</sub> in ocean surface waters at a similar rate (IPCC, 2013). Atmospheric CO<sub>2</sub> is projected to rise to 750–1000 ppm and higher in 2100 (IPCC, 2001) corresponding with a decrease in open ocean pH by 0.3–0.5 units (Caldeira and Wickett, 2005) from the present level of  $\sim 8.1$ . Although this process is of global significance and all parts of the oceans are at risk, there will be regional differences in the degree of acidification (Borges et al., 2005). Thus, to determine the CO<sub>2</sub>-related changes in the oceans, multiple studies in different regions are required. Semi-enclosed coastal regions, such as the Baltic Sea, react with higher changes in pH to CO<sub>2</sub> elevation than open ocean waters due to high freshwater inputs resulting in a reduced buffer capacity (Orr, 2011).

In the Baltic Sea, several studies of CO<sub>2</sub> effects have been undertaken on the organism level of fish (Frommel et al., 2013), zooplankton (Pansch et al., 2012; Vehmaa et al., 2012), macrophytes (Pajusalu et al., 2013), benthic species (Hiebenthal et al., 2013; Stemmer et al., 2013), and filamentous cyanobacteria (Czerny et al., 2009; Eichner et al., 2014; Wannicke et al., 2012). Studies on the impacts of elevated CO<sub>2</sub> at the ecosystem level, however, have thus far been limited to Kiel Bight in the western Baltic Sea (Engel et al., 2014; Rossoll et al., 2013; Schulz and Riebesell, 2013), which may fundamentally differ from other parts of the Baltic Sea.

Next to nitrogen, phosphorus (P) controls the productivity of phytoplankton in the ocean (Karl, 2000; Sanudo-Wilhelmy et al., 2001; Tyrrell, 1999) and is a limiting factor in some regions (Ammerman et al., 2003). The total phosphorus (TP) pool comprises phosphate (PO<sub>4</sub>), dissolved organic phosphorus (DOP), and particulate organic (POP) and inorganic (PIP) phosphorus. There is a continuous transformation of phosphorus between these P species due to their uptake, conversion, and release by organisms as well as by interaction with minerals. While PO<sub>4</sub> is the preferred P-species of phyto- and bacterioplankton, DOP can become an important P source when PO<sub>4</sub> is depleted (Llebot et al., 2010; Lomas et al., 2010). DOP includes nucleic acids, phospholipids, and adenosine triphosphate (ATP; Karl and Björkman, 2002) which are structural and functional components of all living cells, but, can also be released into the surrounding water.



**Figure 1.** The Baltic Sea and the location near the peninsula Hango in the western Gulf of Finland where the mesocosms were deployed.

In general, there is little knowledge on how the P cycle is affected by ocean acidification and how related changes in P availability influence the response of organisms to CO<sub>2</sub> elevation. In CO<sub>2</sub> manipulation experiments, particulate phosphorus dynamics were studied to determine effects on C:P stoichiometry of phytoplankton (Riebesell and Tortell, 2011; Sugie and Yoshimura, 2013) and PO<sub>4</sub> concentration dynamics to estimate its utilization (Bellerby et al., 2008). CO<sub>2</sub> effects on phosphorus pool sizes and PO<sub>4</sub> uptake have so far been studied by Tanaka et al. (2008) in the Raunefjorden, Norway and by Unger et al. (2013) and Endres et al. (2013) in laboratory experiments with cultures of *Nodularia spumigena*. In order to reduce the gap of knowledge, we studied the impact of elevated CO<sub>2</sub> on phosphorus pool sizes, the DOP composition, and PO<sub>4</sub> uptake of a northern Baltic Sea plankton community. These measurements provide important information on potential changes in P cycling under increasing CO<sub>2</sub> levels and contribute to a better understanding of the P cycle in brackish water ecosystems.

## 2 Material and methods

### 2.1 Experimental design and CO<sub>2</sub> manipulation

The study was conducted in the northwestern Gulf of Finland, in the proximity of the Tvärminne Zoological Station (TZS; Fig. 1), between 17 June and 4 August 2012, using the KOSMOS mesocosm system (Riebesell et al., 2013). Nine mesocosms (M1–M9) were moored in the open waters of the Storfjärden (59°51.5' N, 23°15.5' E) at a water depth

of ~30 m. Only six of them were included throughout the whole study period since leakages in the remaining three rendered them unusable. Equipment and deployment procedures are described in detail by Paul et al. (2015b). Briefly, polyurethane enclosure bags of 2 m in diameter and 18.5 m in length were mounted in floating frames and lowered in such a way that ~17 m of each bag was immersed in the water column and ~1.5 m remained above the water surface. Large organisms were excluded from the mesocosms by a 3 mm mesh installed at the top and bottom of the bags before closure. The mesocosms were deployed 10 days prior to CO<sub>2</sub> manipulation to rinse the bags and for full water exchange. Sediment traps were mounted on the lower ends to close them water tight, while the upper ends were raised above the water surface to prevent water entry during wave action. The mesocosms were covered with a dome-shaped roof to prevent nutrient input by birds and potentially significant fresh water input by rain. Salinity gradients were removed by bubbling the mesocosms with compressed air for 3.5 min, so that 5 days before the start of the experiment (day -5) the water body was fully homogeneous.

CO<sub>2</sub> was injected at day 0 and the subsequent 4 days by pumping various quantities of 50 µm-filtered and CO<sub>2</sub>-enriched fjord water into seven of the mesocosms as described by Riebesell et al. (2013). The intended CO<sub>2</sub> and pH gradients were reached after the last treatment on day 4. Details are described in Paul et al. (2015b). For the two untreated (control) mesocosms, only filtered fjord water was added to adjust the water volume to that of the treated mesocosms. To compensate for outgassing, the CO<sub>2</sub> manipulation was similarly repeated in the upper 7 m layer of the mesocosms on day 16.

## 2.2 Sampling

Daily sample collection started 3 days before the first CO<sub>2</sub> injection (day -3). Parallel samples were taken from the surrounding fjord. Sampling over the entire 17 m depth was carried out using an integrating water sampler (IWS HYDROBIOS-KIEL) that was lowered slowly on a cable by hand. The sampling frequency differed depending on the parameter to be observed as shown in the overview by Paul et al. (2015b).

Phosphorus pool parameters and uptake rates were determined every second day, except for dissolved organic phosphorus (DOP) components, which were measured every 4 days. Termination of the measurements varied due to logistical constraints. Thus, total phosphorus (TP) and DOP were sampled only until day 29 whereas other parameters were sampled until day 43.

The collected water was filled in HCl-cleaned polyethylene canisters that had been pre-rinsed with sample water. All containers were stored in the dark. Back on land, subsamples were processed immediately for each P-analysis. The other analyses were carried out within a few hours of sample col-

lection and sample storage in a climate room at in situ temperature.

## 2.3 Analytical methods

### 2.3.1 Temperature, salinity, and carbonate chemistry

Measurements in the fjord and in each mesocosm were conducted using a CTD60M memory probe (Sea and sun technology, Trappenkamp, Germany) lowered from the surface to a depth of 17 m at about 0.3 m s<sup>-1</sup> in the early afternoon (01:30–02:30 p.m.). For these parameters, depth-integrated mean values are presented here.

The carbonate system is described in detail in Paul et al. (2015b). The pHT (total scale) was determined using a spectrophotometric method (Dickson et al., 2007) on a Cary 100 (Varian) and the dye m-cresol as indicator. Extinction was measured at 578 (E1) and 434 nm (E2) in a 10 cm cuvette. The pH was calculated from the ratio of E1 and E2 (Clayton and Byrne, 1993).

DIC was measured using a coulometric AIRICA system (MARIANDA, Kiel) measuring the infrared absorption after N<sub>2</sub> purging of the sample and calibration with certified reference material (CRM; A. Dickson, University of California, San Diego).

The *f*CO<sub>2</sub> was calculated from DIC, pHT, salinity and using the stoichiometric equilibrium constant for carbonic acid of Mehrbach et al. (1973) as refitted by Lueker et al. (2000).

### 2.3.2 Chlorophyll and inorganic nutrients

Subsamples of 500 mL were filtered onto GF/F-filters. Chl *a* was extracted in acetone (90 %) in plastic vials by homogenisation of the filters for 5 min in a cell mill using glass beads. After centrifugation (10 min, 800 × *g*, 4 °C) the supernatant was analysed on a fluorometer (TURNER 10-AU) at an excitation of 450 nm and an emission of 670 nm to determine Chl *a* concentrations (Jeffrey and Welschmeyer, 1997).

A segmented continuous-flow analyzer coupled with a liquid-waveguide capillary flow-cell (LWCC) of 2 m length was used to determine phosphate (PO<sub>4</sub>) and the sum of nitrite and nitrate (NO<sub>2</sub> + NO<sub>3</sub>) at nanomolar precision (Patey et al., 2008). The PO<sub>4</sub> determination was based on the molybdenum blue method of Murphy and Riley (1962), and NO<sub>3</sub>+NO<sub>2</sub> on the method of Morris and Riley (1963). PO<sub>4</sub> concentrations from the same subsample were also measured manually using a 5 cm cuvette (Grasshoff et al., 1983). In most of the samplings PO<sub>4</sub> data obtained from both methods did not differ significantly (paired *t* test: *p* = 0.262, *t* = 1.127, *n* = 109).

### 2.3.3 Dissolved organic phosphorus (DOP)

For the determination of DOP, duplicate 40 mL subsamples were filtered through pre-combusted (6 h, 450 °C) glass fiber filters (Whatman GF/F) and stored in 50 mL vials (Falcon) at

−20 °C until further processing. The thawed samples were oxidized in a microwave (MARSHpress, CEM, Matthews, USA; Johnes and Heathwaite, 1992) after the addition of potassium peroxydisulfate in an alkaline medium (Bhaya et al., 2000). The P concentration, measured as PO<sub>4</sub> in a 10 cm cuvette, represents the total dissolved phosphorus (DP) concentration. DOP was calculated as the difference between the DP concentrations in the filtered and digested samples and the corresponding PO<sub>4</sub> concentration analysed as described above.

### 2.3.4 Dissolved organic phosphorus compounds

For all analysed components, subsamples were pre-filtered through pre-combusted (6 h, 450 °C) filters (Whatman GF/F) to remove larger particles followed by filtration through 0.2 µm cellulose acetate filters to remove picoplankton. Subsamples were prepared for storage according to the specific method used for each compound. After the analyses, the phosphorus content of measured DOP compounds was summed and the amount subtracted from the total DOP concentration. The difference is defined as the uncharacterized DOP.

### Dissolved ATP

The method of (Björkman and Karl, 2001) adapted to Baltic Sea conditions (Unger et al., 2013) was used to determine dissolved adenosine triphosphate (dATP). An Mg(OH)<sub>2</sub> precipitate, including the co-precipitated nucleotides, was obtained by treating 200 mL of the filtrate with 2 mL of 1 M NaOH (1 % *v/v*). The precipitate was allowed to settle overnight and then centrifuged at 1000 g for 15 min. The supernatant was discarded and the precipitate was transferred into 50 mL Falcon tubes, centrifuged again (1.5 h, 1680 × *g*). The resulting pellet was dissolved by drop-wise addition of 5 M HCl. The samples were frozen at −20 °C until further processing. The pH of the thawed samples was adjusted to 7.2 by the addition of TRIS buffer (pH 7.4, 20 mM). The final volume was recorded. The dATP concentrations were measured in triplicate using the firefly bioluminescence assay and a Sirius luminometer (Berthold Detection Systems Pforzheim, Germany), as described by Unger et al. (2013). Standard concentrations were prepared as described above, using aged Baltic Sea water and six ATP concentrations (adenosine 50-triphosphate disodium salt hydrate, Sigma-Aldrich, A2383) ranging from 1 to 20 nmol L<sup>−1</sup>. The detection limit of the bioluminescence assay was 2.5 nmol L<sup>−1</sup>. The fluorescence slope of the standard concentrations was used to calculate dATP concentrations, correcting for the final sample volume. The P-content of the dATP (dATP-P) was calculated by assuming that 1 mol of ATP is equivalent to 3 mol P.

### Dissolved phospholipids

The phosphate content of the dissolved phospholipids (PL-P) was analysed using a modified method of Suzumura and In-gall (2001, 2004). Briefly, 400 mL subsamples of the filtrate were stored at −20 °C until further processing. The samples were then thawed in a water bath at 30 °C and extracted twice with 100 mL of chloroform. The chloroform phase was collected, concentrated to 5 mL in a rotary evaporator (Heidolph Instruments, Schwabach, Germany), and then transferred into microwave tubes. The chloroform was completely evaporated by incubating the tubes in a 60 °C water bath overnight. After the addition of 20 mL of deionized water (Milli-Q, Millipore), the samples were digested with potassium peroxydisulfate in alkaline medium and microwaved as described for the DOP analysis. Six standard concentrations of phospholipids, ranging from 0 to 125 µg L<sup>−1</sup>, were prepared by adding the respective amounts of a stock solution containing 5 mg of 1-phosphatidyl-dl-glycerol sodium salt (PG, Sigma Aldrich, P8318) mL<sup>−1</sup> to the aged seawater. The detection limit was 0.8 nmol L<sup>−1</sup>. The blanks contained only chloroform and were processed as for the samples.

### Dissolved DNA and RNA

Dissolved DNA and RNA (dDNA and dRNA) concentrations were determined according to Karl and Bailiff (1989) and as described by Unger et al. (2013). For each sample, 200 mL of the filtrate was gently mixed with the same volume of ethylene-diamine-tetracetic acid (EDTA, 0.1 M, pH 9.3, Merck, 1.08454) and 4 mL of cetyltrimethyl-ammonium bromide (CTAB, Sigma-Aldrich, H5882) and stored frozen at −20 °C for at least 24 h. After thawing the samples, the precipitate was collected onto combusted (450 °C, 6 h) glass fiber filters (25 mm, GF/F Whatman), placed into annealed vials, and stored frozen at −80 °C until further analysis.

DNA concentrations were measured using a fluorescence-spectrophotometer (Hitachi F 2000), and RNA concentrations using a dual-beam UV/VIS-spectrophotometer U3010 (Hitachi).

Coupled standards (DNA + RNA) containing 1–10 µg DNA (Sigma Aldrich, D3779) L<sup>−1</sup> and 20–120 µg RNA (Sigma Aldrich, R1753) L<sup>−1</sup> were prepared in aged seawater as described above. A reagent blank served as the reference and aged seawater as the background control. The P-contents of the DNA and RNA were calculated by multiplying the measured values by a factor of 2.06 nmol P per µg dDNA and 2.55 nmol P per µg dRNA. The latter values were determined by the microwave digestion of standard substrates.

### 2.3.5 Particulate organic phosphorus, carbon, and nitrogen

Particulate phosphorus (PP) was analysed using two methods in parallel. In the “aqueous method”, 40 mL of unfil-

tered subsamples were frozen at  $-20^{\circ}\text{C}$  and analysed as described for DOP using the potassium peroxydisulfate digestion (Grasshoff et al., 1983). The measured PO<sub>4</sub> concentration represents total phosphorus (TP). PP is the difference between the TP concentration in the unfiltered digested sample and the sum of DOP+PO<sub>4</sub>. In the “filter-method”, 500 mL subsamples were filtered onto pre-combusted GF/F-filters that were then placed into Schott bottles containing 40 mL of deionised water. PP was digested to PO<sub>4</sub> by the addition of oxidizing decomposition reagent (Oxisolv<sup>®</sup>, Merck) followed by heating in a pressure cooker for 30 min. The PO<sub>4</sub> concentrations of the cooled samples were determined spectrophotometrically according to Grasshoff et al. (1983). Paired *t* test revealed significant differences between two methods; however, the difference between the means of the filter method and of the aqueous method ( $0.19 \pm 0.03 \text{ mmol L}^{-1}$  and  $0.1 \pm 0.04 \text{ } \mu\text{mol L}^{-1}$ , respectively) were near the detection limit ( $0.02 \text{ } \mu\text{mol L}^{-1}$ ) of the methods. Thus, solely the mean values obtained from both measurements are used in the following.

Particulate carbon (PC) and nitrogen (PN) were analysed by filtering 500 mL samples onto pre-combusted ( $450^{\circ}\text{C}$ , 6 h) glass fiber filters (Whatman GF/F), which were then stored frozen at  $-20^{\circ}\text{C}$ . PC and PN concentrations were measured by flash combustion of the dried ( $60^{\circ}\text{C}$ ) filters using a EuroEA elemental analyser coupled with a Conflo II interface to a Finnigan Delta<sup>Plus</sup> mass spectrometer and included organic and inorganic matter.

### 2.3.6 Phosphate and ATP uptake

PO<sub>4</sub> uptake was measured by addition of radioactively labelled phosphate [<sup>33</sup>P]PO<sub>4</sub> (specific activity of  $111 \text{ TBq mmol}^{-1}$ , Hartmann Analytic GmbH) at concentrations of  $50 \text{ pmol L}^{-1}$  to 50 mL subsamples, which were then incubated under laboratory light and the in situ temperatures for  $\sim 2 \text{ h}$ . For each mesocosm, three parallel samples and a blank were prepared. The blank was obtained by the addition of formaldehyde (1 % final concentration) 10 min before radiotracer addition, in order to poison the samples. At defined time intervals within the incubation, 5 mL subsamples were taken from each of the parallel samples and filtered onto polycarbonate (PC) filters pre-soaked with a cold 20 mM PO<sub>4</sub> solution to prevent non-specific [<sup>33</sup>P]PO<sub>4</sub> binding. The filters were rinsed with 5 times 1 mL of particle-free bay water and placed in 6 mL scintillation vials. Scintillation liquid (4 mL IrgaSafe; Perkin Elmer) was added and the contents of the vials were mixed using a vortex mixer. After allowing the samples to stand for at least 2 h, the radioactivity on the filters was counted in a Perkin Elmer scintillation counter. PC-filters of 0.2  $\mu\text{m}$  and 3  $\mu\text{m}$  pore sizes (Whatman and Millipore, respectively) were used to determine uptake by the whole plankton community and the size fraction  $> 3 \text{ } \mu\text{m}$ , respectively. Picoplankton uptake was calculated as the difference between the activity on the 0.2 and 3  $\mu\text{m}$  filters.

[ $\gamma$ -<sup>33</sup>P]ATP (specific activity of  $111 \text{ TBq mmol}^{-1}$ , Hartmann Analytic GmbH) was added to triplicate 10 mL samples and a blank, each in a 20 mL vial, at a concentration of  $50 \text{ pmol L}^{-1}$ . The samples were incubated in the dark at the in situ temperature for 1 h. The uptake was stopped by addition of 200  $\mu\text{L}$  of a cold 20 mM ATP solution to the samples, which were then filtered and processed as described for the PO<sub>4</sub> uptake measurements.

### 2.3.7 Bacterial production (BPP)

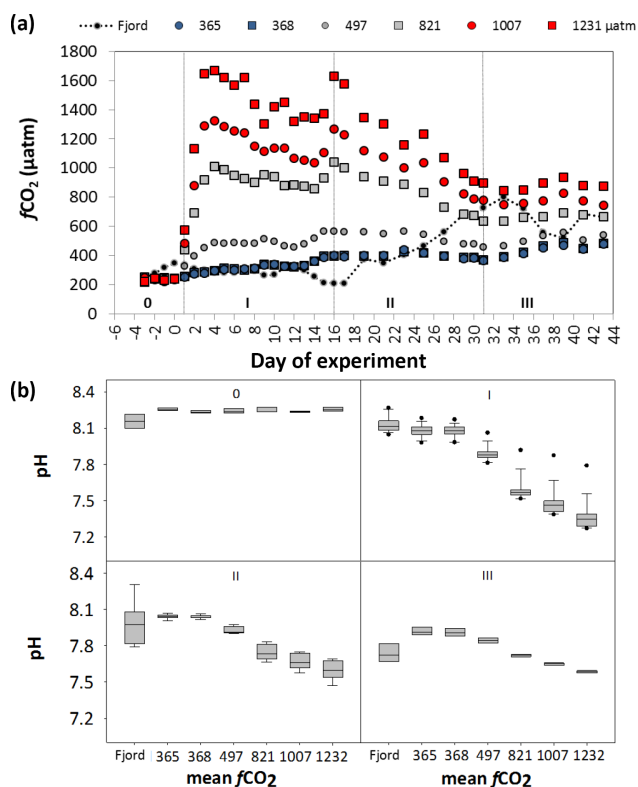
Rates of bacterial protein production (BPP) were determined by incorporation of [<sup>14</sup>C]-leucine (<sup>14</sup>C-Leu, Simon and Azam, 1989) according to Grossart et al. (2006). Triplicates and a formalin-killed control were incubated with <sup>14</sup>C-Leu ( $7.9 \text{ GBq mmol}^{-1}$ ; Hartmann Analytic GmbH, Germany) at a final concentration of  $165 \text{ nmol L}^{-1}$ , which ensured saturation of uptake systems of both free and particle-associated bacteria. Incubation was performed in the dark at in situ temperature (between  $7.8$  and  $15.8^{\circ}\text{C}$ ) for 1.5 h. After fixation with 2 % formalin, samples were filtered onto 5.0  $\mu\text{m}$  (attached) nitrocellulose filters (Sartorius, Germany) and extracted with ice-cold 5 % trichloroacetic acid (TCA) for 5 min. Thereafter, filters were rinsed twice with ice-cold 5 % TCA, once with ethanol (96 % *v/v*), and dissolved with ethylacetate for measurement by liquid scintillation counting. Afterwards the collected filtrate was filtered on 0.2  $\mu\text{m}$  (free-living) nitrocellulose filters (Sartorius, Germany) and processed in the same way as the 5.0  $\mu\text{m}$  filters. Standard deviation of triplicate measurements was usually  $< 15 \%$ . The sum of both fractions (free-living bacteria and attached bacteria) is referred to total BPP. The amount of incorporated <sup>14</sup>C-Leu was converted into BPP by using an intracellular isotope dilution factor of 2 (Simon and Rosenstock, 1992). A conversion factor of 0.86 was used to convert the protein produced into carbon (Simon and Azam, 1989).

## 2.4 Statistical analyses

The Grubbs test, done online ([graphpad.com/quickcalcs/Grubbs1.cfm](http://graphpad.com/quickcalcs/Grubbs1.cfm)) was applied to identify outliers in all data sets. The outliers were removed from further statistical analyses.

Spearman Rank correlations were carried out to describe the relationship between the development of the parameters over time in the mesocosms and in the fjord using Statistica 6 software.

Short-term CO<sub>2</sub> effects on PP concentrations at days 0–2 and 23–43 between the CO<sub>2</sub> treatments were verified with an ANCOVA analysis using the SPSS software. The “days” were treated as a covariate interacting with the treatments. Paired *t* test was applied to check the differences in PO<sub>4</sub> concentrations between the treatments.



**Figure 2.** (a)  $f\text{CO}_2$  values in the mesocosms and in the fjord throughout the experiment. Small black dots show the  $f\text{CO}_2$  in the ambient fjord water. Treatment of the mesocosms with CO<sub>2</sub> saturated fjord water at the beginning of the experiment (days 0–4) created different  $f\text{CO}_2$  levels in the mesocosms: blue symbols represent the untreated mesocosms, grey the intermediate, and red the high CO<sub>2</sub> treated mesocosms. The treatment was repeated at day 16. (b) Corresponding pH ranges in the mesocosms during the four phases. Despite decreasing trends over time, a gradient between the mesocosms was kept over the whole period.

### 3 Results

#### 3.1 Development in the mesocosms

##### 3.1.1 CO<sub>2</sub>, pH, temperature and salinity

The different mesocosms were characterized based on their averaged  $f\text{CO}_2$  and pH values from day 1 until day 43 (Fig. 2a, b):

- M1 365 μatm  $f\text{CO}_2$ , pH 8.08
- M5 368 μatm  $f\text{CO}_2$ , pH 8.07
- M7 497 μatm  $f\text{CO}_2$ , pH 7.95
- M6 821 μatm  $f\text{CO}_2$ , pH 7.74
- M3 1007 μatm  $f\text{CO}_2$ , pH 7.66
- M8 1231 μatm  $f\text{CO}_2$ , pH 7.58

M1 and M5 were the untreated mesocosms and served as controls.

Temperature development in the mesocosms closely followed that in the fjord ranging from 7.82 to 15.86 °C. Based

**Table 1.** Minimum, maximum and mean values of hydrographical parameters and  $f\text{CO}_2$  for the different phases in the fjord. Temperatures in the mesocosms were identical with those in surrounding fjord water.

phase	min	max	mean
water temperature (°C)			
0	7.82	8.71	8.20
I	9.66	15.86	12.27
II	7.89	14.79	11.68
III	8.35	12.61	10.83
salinity			
0	5.72	5.85	5.78
I	5.46	5.85	5.65
II	5.67	6.04	5.82
III	5.9	6.05	5.98
pH			
0	8.09	8.23	8.16
I	8.11	8.30	8.17
II	7.81	8.30	8.00
III	7.75	7.93	7.83
$f\text{CO}_2$ (μatm)			
0	250	347	298
I	207	336	283
II	208	679	465
III	521	800	668

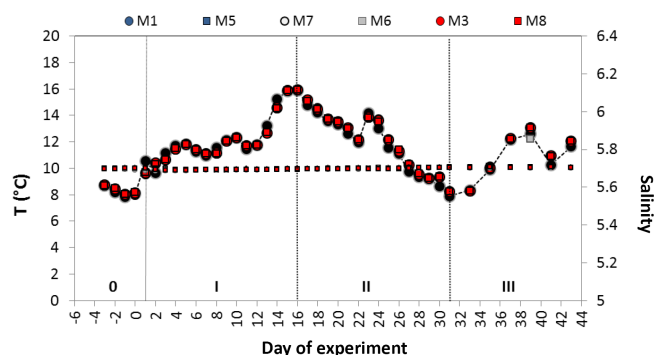
on this (compare Paul et al., 2015b for details), the experiment was divided into four phases (Fig. 3): phase 0: day –3 to day 0; phase I: days 1–16, phase II: days 17–30 and phase III: day 31 until the end of the measurements. Temperature dropped from 8.71 to 7.82 °C in phase 0 and rose from 8.07 °C at the start of phase I to the maximum of 15.86 °C by the end of this phase. During phase II, the temperature decreased to 7.89 °C interrupted by a short reversal on days 22 and 23. During phase III, the temperature increased to 12.61 °C (Table 1).

Salinity ( $5.69 \pm 0.01$ ) remained relatively stable in all mesocosms throughout the entire experimental period (Fig. 3).

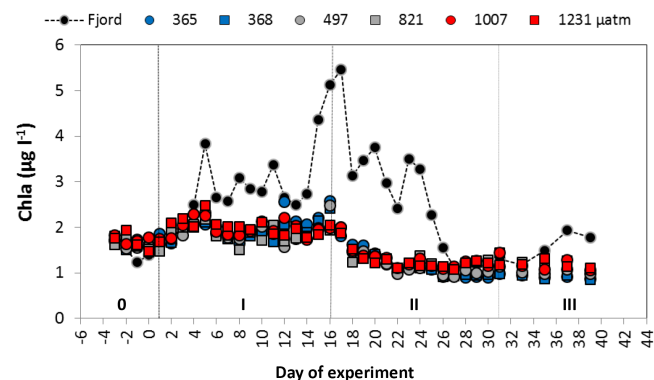
##### 3.1.2 Phytoplankton biomass

Chlorophyll *a* (Chl *a*) reached maximum concentrations of  $2.06\text{--}2.48 \mu\text{g L}^{-1}$  at day 5 (Fig. 4). Average concentrations of  $1.94 \pm 0.23 \mu\text{g L}^{-1}$  in phase I exceeded those in phases II and III when Chl *a* decreased to a mean of  $1.08 \pm 0.16 \mu\text{g L}^{-1}$ . The increase in Chl *a* in the high CO<sub>2</sub> mesocosms by  $0.27 \mu\text{g L}^{-1}$  in phase III was marginal for Baltic Sea summer conditions. According to Paul et al. (2015b), this represents an increase of 24 % which is a significant difference compared to the controls.





**Figure 3.** Temperature and salinity averaged over the 17 m surface layer of the mesocosms and the fjord. The data were obtained from daily CTD casts. Large symbols represent temperature and the small symbols salinity. Fjord water is shown as black dots with broken line while blue symbols denote untreated, grey intermediate and red high  $f\text{CO}_2$  levels in the mesocosms. According to the temperature regime, the experimental period can be divided into four phases (phases 0, I, II and III).



**Figure 4.** Chl *a* concentrations in fjord water and in the mesocosms with different  $f\text{CO}_2$  conditions. The development over time can be divided into three phases as well. Blue represents untreated, grey intermediate, and red highly treated  $f\text{CO}_2$  levels. Black dots with dotted line are the Chl *a* concentrations in the fjord water.

We observed a significant relationship between Chl *a* and  $\text{PO}_4$  in the untreated and intermediate treated mesocosms that diminished with increasing  $f\text{CO}_2$  as indicated by lower  $p$  values. The statistical significance was lost in the highest  $f\text{CO}_2$  mesocosms (Table 2).

### 3.1.3 Phosphorus pools

Total phosphorus (TP) concentrations in the mesocosms ranged between  $0.49$  and  $0.68 \mu\text{mol L}^{-1}$  (Fig. 5a) during the experiment without statistically significant differences between the CO<sub>2</sub> treatments. Shortly after the bags were closed, the decline in TP concentrations began and continued until the beginning of phase II. On average, TP concentrations decreased from  $0.63 \pm 0.02$  on day  $-3$  to  $0.51 \pm 0.01 \mu\text{mol L}^{-1}$

on day 21. Thereafter, the mean TP remained constant at  $0.54 \pm 0.03 \mu\text{mol L}^{-1}$  until the end of the measurements. Thus, the loss of phosphorus ( $116 \pm 34 \text{ nmol L}^{-1}$ ) from the 17 m layer during the 29-day measurement period was calculated to be  $4.0 \text{ nmol L}^{-1} \text{ day}^{-1}$ . The decline in TP can be explained by loss through sedimentation of PP (Paul et al., 2015b).

Particulate phosphorus (PP) concentrations varied from  $0.10$  to  $0.23 \mu\text{mol L}^{-1}$  in all CO<sub>2</sub> treatments (Figs. 5b, 6). We expected that the decrease in TP was reflected in PP. However, parallel changes occurred only periodically. PP concentrations increased during the first 5 days after the bags were closed. This increase was stimulated by CO<sub>2</sub> additions from day 0 to day 2 (ANCOVA:  $p = 0.004$ ,  $F = 20.811$ ; Fig. 7a). Subsequently, PP declined in parallel with TP until day 21, albeit with a lower amount. Averaged over all mesocosms, TP decreased by  $0.12 \pm 0.03 \mu\text{mol L}^{-1}$ , whereas PP declined only by  $0.06 \pm 0.01 \mu\text{mol L}^{-1}$  during this period. From day 23 until the end of the measurements, PP remained at relatively constant concentrations; however, PP concentrations in the high CO<sub>2</sub> treated mesocosms exceeded those in the other mesocosms significantly (ANCOVA:  $p < 0.0001$ ,  $F = 11.99$ ; Figs. 5b, 7). PP developed in parallel with PC. The two parameters were positively correlated in the untreated and the intermediate CO<sub>2</sub> treatments, but not in the high CO<sub>2</sub> treatments (Table 2). Figures 3 and 6b show that the increase in Chl *a* was delayed by 2–3 days compared to the increase in PP during the first growth event. A correlation between PP and Chl *a* was detected only for the untreated mesocosms (Table 2).

Dissolved organic phosphorus (DOP) concentrations in the mesocosms ranged between  $0.18$  and  $0.36 \mu\text{mol L}^{-1}$  constituting 32–71 % of the TP pool (Fig. 5c). DOP did not change significantly in response to the CO<sub>2</sub> perturbations, and were similar to the concentrations in fjord water. Concentrations of  $\geq 0.3 \mu\text{mol L}^{-1}$  were measured on days 6 and 7 (phase I) and on day 23 (phase II); the high DOP value in the intermediate CO<sub>2</sub> treatment at day 19 was identified as an outlier according to Grubbs test (Fig. 5c).

In phase I, DOP initially increased in parallel with Chl *a* and BPP but reached its maximum 1–2 days later, after which it decreased only marginally until the end of this phase, independent of changes in BPP and Chl *a* (Fig. 8c, d). In phase II, the peak conformed to that of BPP. DOP correlated with temperature only in the high  $f\text{CO}_2$  mesocosms (Table 2). In addition, the composition of DOP did not change with increasing CO<sub>2</sub> (Fig. 10). The sum of RNA ( $\sim 47$  %) plus the unidentified fraction constituted 98–99 % of the DOP pool whereas the other measured compounds contributed only 1–2 % (Table 3).

Phosphate ( $\text{PO}_4$ ) concentrations ranged between  $0.06$  and  $0.21 \mu\text{mol L}^{-1}$ , with deviations between the mesocosms only in nanomolar range. The mean contribution of  $\text{PO}_4$  to TP was  $25 \pm 6$  %, which was the lowest among all TP fractions (Fig. 6). From the start of the measurements to day

**Table 2.** Mesocosms in which the Spearman Rank correlation between P-pools or uptake rates and other parameters was significant. The relationship of PP with TP and Chl *a* was significant only in the untreated mesocosms while the correlation to PC was also significant in the mesocosms with intermediate CO<sub>2</sub> levels. DOP was related to temperature only in the high CO<sub>2</sub> treatments. Under high *f*CO<sub>2</sub> conditions, the PO<sub>4</sub> uptake in the size fraction > 3 µm correlated with Chl *a* and the P content of phytoplankton.

Relationship between	<i>f</i> CO <sub>2</sub> (µatm)	Significant responses		
		<i>r</i>	<i>p</i>	<i>n</i>
PP–TP	365	0.599	0.008	18
	368	0.515	0.029	18
PP–Chl <i>a</i>	365	0.479	0.0130	25
	368	0.584	0.0022	25
PO <sub>4</sub> –Chl <i>a</i>	365	−0.832	<0.0001	21
	368	−0.756	0.0011	20
	497	−0.674	0.0008	21
	821	−0.524	0.0147	21
	1007	−0.634	0.0027	20
PP–PC	365	0.542	0.0061	24
	368	0.625	0.0011	24
	497	0.404	0.0490	24
	821	0.551	0.0052	24
DOP–temperature	1007	0.488	0.0470	17
	1231	0.525	0.0310	17
PO <sub>4</sub> uptake > 3 µm–Chl <i>a</i>	497	0.743	0.0056	12
	821	0.674	0.0081	14
	1231	0.476	0.0310	14
PO <sub>4</sub> uptake > 3 µm–POP/Chl <i>a</i>	497	−0.601	0.0380	12
	821	−0.631	0.0160	14
	1231	−0.626	0.0165	14

**Table 3.** Contribution of different phosphorus components to DOP in the mesocosms and in the fjord.

<i>f</i> CO <sub>2</sub> (µatm)	Contribution to DOP (%)					
	ATP-P	PL-P	DNA-P	RNA-P	sum	unidentified P
Fjord	0.7	0.7	0.04	69.4	70.84	29.16
365	0.7	0.5	0.03	44.1	45.33	54.67
368	0.6	0.5	0.03	46.9	48.03	51.97
497	0.6	0.4	0.04	49.5	50.54	49.46
821	0.6	0.4	0.03	41.8	42.83	57.17
1003	0.8	0.4	0.04	60.1	61.34	38.66
1231	0.5	0.4	0.03	48.6	49.53	50.47

13, PO<sub>4</sub> declined by 0.06 µmol L<sup>−1</sup> (or 3.5 nmol L<sup>−1</sup> day<sup>−1</sup>) from initial values of 0.16 ± 0.01 µmol L<sup>−1</sup> (Fig. 5d). Subsequently, concentrations increased again, by an average of 2.6 nmol L<sup>−1</sup> day<sup>−1</sup>, until the end of the experiment. There were no significant differences between CO<sub>2</sub> treatments until day 23, when high CO<sub>2</sub> concentrations led to slightly lower PO<sub>4</sub> concentrations (Fig. 5d). Afterwards, PO<sub>4</sub> concentrations in the high *f*CO<sub>2</sub> mesocosms were significantly lower than those in the untreated mesocosms (*t* = 6.51, *p* = 0.0003). This observation is in accordance with the dy-

namics of PP and Chl *a* concentrations, which were significantly elevated in the high CO<sub>2</sub> treatments. Thus, the transformation of PO<sub>4</sub> to POP via stimulated biomass formation may have been promoted under high CO<sub>2</sub> conditions in phase III.

Since PO<sub>4</sub> was never fully exhausted, phosphorus limitation of phyto- and bacterioplankton can be excluded. This interpretation is supported by the PC : PP ratios, which varied between 84.4 and 161.1 in all treatments (Paul et al., 2015b) deviating only slightly from the Redfield ratio.

### 3.1.4 Uptake of PO<sub>4</sub> and ATP

PO<sub>4</sub> turnover times of 1.5–8.4 days (mean 4.0 ± 1.2 days, *n* = 112) in all mesocosms indicated no dependency on the CO<sub>2</sub> treatment (Fig. 9a). Gross PO<sub>4</sub> uptake rates were in the range of 0.6–3.9 nmol L<sup>−1</sup> h<sup>−1</sup> (mean 1.7 ± 0.6 nmol L<sup>−1</sup> h<sup>−1</sup>, *n* = 112), or 14.3–94.4 nmol L<sup>−1</sup> day<sup>−1</sup> (mean 41.3 ± 13.8 nmol L<sup>−1</sup> day<sup>−1</sup>; Fig. 9b, Table 4). The rates were highest on days 4 and 9 (phase I) and decreased thereafter until day 15, followed by an increase to a mean maximum rate of 2.3 ± 0.5 nmol L<sup>−1</sup> h<sup>−1</sup> (*n* = 6) at day 27. The size fraction < 3 µm was responsible for 59.1 to



**Table 4.** PO<sub>4</sub>- and ATP uptake rates in the fjord and in the mesocosms. Minimum, maximum and mean values as well as the contribution of the size fraction < 3 µm to the total activity are given for the whole period of investigation (each:  $n = 16$  for PO<sub>4</sub> and  $n = 6$  for ATP uptake).

$f\text{CO}_2$	Total PO <sub>4</sub> uptake ( $\text{n mol l}^{-1} \text{ h}^{-1}$ )			portion (%) < 3 µm	Total ATP-P uptake ( $\text{n mol l}^{-1} \text{ h}^{-1}$ )			portion (%) < 3 µm
	min	max	mean		min	max	mean	
Fjord	0.87	2.81	$1.63 \pm 0.58$	$76 \pm 15$	0.04	0.51	$0.26 \pm 0.15$	$92 \pm 5$
365	0.82	3.89	$1.67 \pm 0.82$	$81 \pm 11$	0.14	1.08	$0.43 \pm 0.33$	$96 \pm 2$
368	0.65	2.74	$1.61 \pm 0.58$	$86 \pm 7$	0.16	0.97	$0.47 \pm 0.27$	$96 \pm 2$
497	0.61	3.03	$1.52 \pm 0.59$	$86 \pm 6$	0.20	1.07	$0.54 \pm 0.28$	$96 \pm 2$
821	0.91	2.83	$1.60 \pm 0.59$	$88 \pm 8$	0.14	0.71	$0.36 \pm 0.21$	$97 \pm 2$
1003	0.67	3.79	$1.73 \pm 0.85$	$86 \pm 6$	0.22	0.69	$0.39 \pm 0.15$	$97 \pm 1$
1231	0.87	2.23	$1.53 \pm 0.43$	$87 \pm 6$	0.17	0.67	$0.44 \pm 0.17$	$97 \pm 2$

98.4 % of the total PO<sub>4</sub> uptake (mean  $86.5 \pm 7.6$  %) whereas the size fraction > 3 µm accounted for only 1.6–40.9 % (mean  $13.5 \pm 7.4$  %). Thus, PO<sub>4</sub> was taken up mainly by picoplankton. However, only the uptake rate by the size fraction > 3 µm was positively related to Chl *a* and inversely related to the P content of the biomass (Table 2). Thus, the PO<sub>4</sub> uptake was obviously stimulated when the phytoplankton biomass increased and the cellular P decreased simultaneously. The relationship between PO<sub>4</sub> uptake by this fraction and Chl *a* became evident only in the CO<sub>2</sub>-amended conditions indicating that the interaction between P uptake, cellular P-content and growth of phytoplankton was stimulated under elevated CO<sub>2</sub> conditions.

ATP turnover times of 0.2 to 3.6 days (mean  $0.94 \pm 0.74$  days,  $n = 90$ ) were much shorter than the PO<sub>4</sub> turnover times and did not vary between the treatments (Fig. 9c). Between 0.05 and 0.36 nmol ATP L<sup>-1</sup> h<sup>-1</sup> (mean  $0.14 \pm 0.08$  nmol L<sup>-1</sup> h<sup>-1</sup>,  $n = 36$ ) were hydrolysed, corresponding to a P supply of 0.14 and 1.08 nmol L<sup>-1</sup> h<sup>-1</sup> (mean  $0.44 \pm 0.25$  nmol L<sup>-1</sup> h<sup>-1</sup>,  $n = 36$ ). Thus, phosphorus additionally supplied from ATP accounted for ~25 % of that provided by PO<sub>4</sub>. The picoplankton size fraction (< 3 µm) was responsible for 90–99 % of ATP uptake, with only a marginal portion (1.6–9.5 %) attributable to the phytoplankton fraction > 3 µm (Table 4).

### 3.2 Hydrography and pool sizes in the fjord

Large variations in  $f\text{CO}_2$  and pH occurred in fjord water during the period of investigation (Table 1). The relationship of  $f\text{CO}_2$  with temperature and salinity indicated that the CO<sub>2</sub> conditions were influenced predominantly by changes in the water masses, specifically by upwelling which affected both the relationship of  $f\text{CO}_2$  with PO<sub>4</sub> and probably the correlation of  $f\text{CO}_2$  with Chl *a* and PC (Table 2).  $f\text{CO}_2$  ranged from 207 µatm (Fig. 2a) at days 12–16 when temperatures were highest to 800 µatm at day 33 when deep water input occurred which was indicated by pH below 7.75.

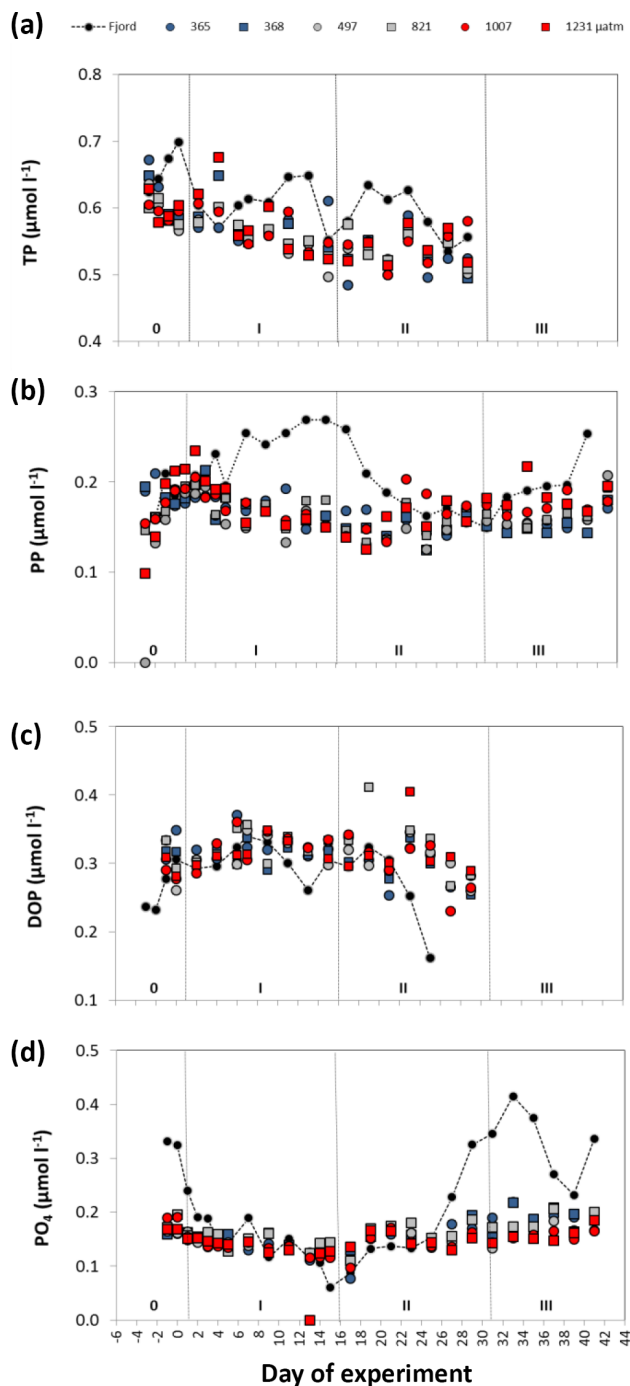
Chl *a* concentrations were between 1.12 and 5.46 µg L<sup>-1</sup> (mean  $2.29 \pm 1.11$  µg L<sup>-1</sup>;  $n = 38$ ), with distinct phases cor-

relating with temperature, salinity and pH. However, the Chl *a* maximum occurred at the beginning of phase II, which was 1–2 days after the maximum temperature. Shortly thereafter, Chl *a* decreased to its lowest level before it increased again, albeit only marginally to 1.93 µg L<sup>-1</sup> during phase III (Fig. 4).

TP concentrations from day -3 until day 29 ranged between 0.54 and 0.70 µmol L<sup>-1</sup> (mean  $0.61 \pm 0.04$  µmol L<sup>-1</sup>;  $n = 19$ ; Figs. 5a, 6). With a general decreasing tendency, TP undulated with a frequency of about 10 days in the period of phases 0 to the first half of phase I and of 6 days in the second half of phase I to II. For the period under investigation, the TP fractions had the following characteristics.

PP concentrations varied from 0.13 to 0.30 µmol L<sup>-1</sup> (mean  $0.20 \pm 0.04$  µmol L<sup>-1</sup>;  $n = 29$ ), thus accounting for 23.4–51.8 % (mean  $34.7 \pm 7.9$  %;  $n = 19$ ) of the TP pool. The development of PP over time did not follow that of TP (Fig. 5b). PP concentrations were highest between days 8 and 19, when the accumulation of PP in the biomass was reflected in declining C : P ratios from 180 to 107 (Paul et al., 2015b) and thereafter remained at the low ratio until the end of the measurements. The PP increase in phase III occurred in parallel to Chl *a* and to the PO<sub>4</sub> decrease (Fig. 6). Thus PO<sub>4</sub> was transformed into PP via biomass production. The calculated P content of phytoplankton was 0.05–0.15 (mean 0.1) µmol PP (µg Chl *a*)<sup>-1</sup>.

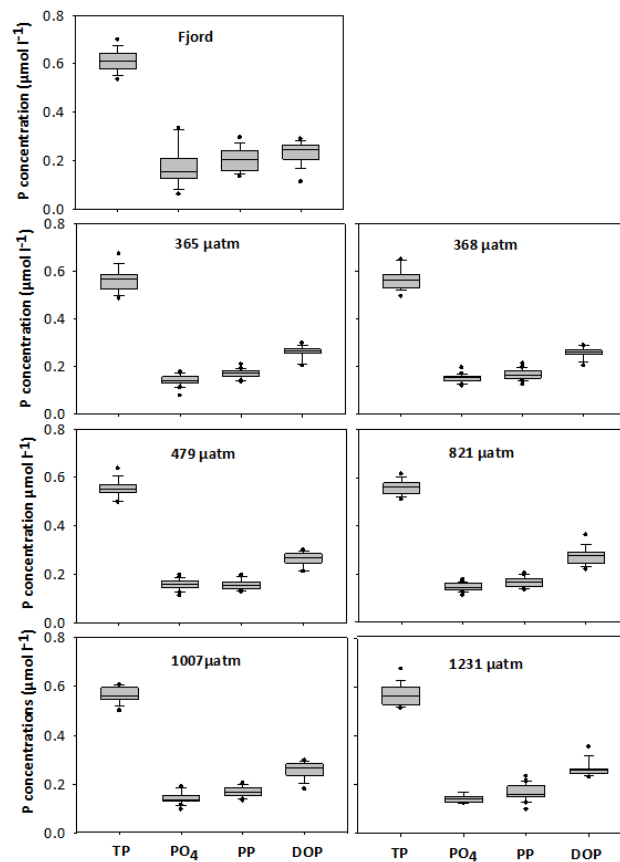
DOP substantially contributed (26–45 %) to the TP pool (Fig. 6). Concentrations ranged between 0.19 and 0.29 µmol L<sup>-1</sup> (mean  $0.24 \pm 0.03$  µmol L<sup>-1</sup>;  $n = 17$ ), with high concentrations occurring in parallel to those of TP in phases I and II (Fig. 5c). The very low DOP value of 0.11 µmol L<sup>-1</sup>, on day 29, was an outlier (Grubbs test). For the whole study period, DOP concentrations correlated positively with PP ( $p = 0.034$ ,  $n = 17$ ) and inversely with PO<sub>4</sub> concentrations ( $p = 0.005$ ,  $n = 17$ ). A similar behaviour between DOP and Chl *a* was restricted to phases 0 and I, whereas the relationship was inverse in phase II (Fig. 8b). As shown in Figure 8a, the DOP and BPP levels alternated with the same rhythm, but inversely, in phases 0 and I and changed to a parallel development in phase II. Statistical analysis was



**Figure 5.** (a–d) Development of total phosphorus (TP) and the three measured P-fractions in fjord water (black dots with dotted line) and in the mesocosms over time. Blue represents untreated, grey intermediate and red high  $f\text{CO}_2$  treatment levels.

not feasible because DOP and BPP were not always sampled on the same day and only very few data pairs were available.

$\text{PO}_4$  concentrations ranged between 0.06 and  $0.41 \mu\text{mol L}^{-1}$  (mean  $0.21 \pm 0.09 \mu\text{mol L}^{-1}$ ,  $n = 21$ ), thus comprising  $24.3 \pm 11.2\%$  ( $n = 21$ ) of the TP pool

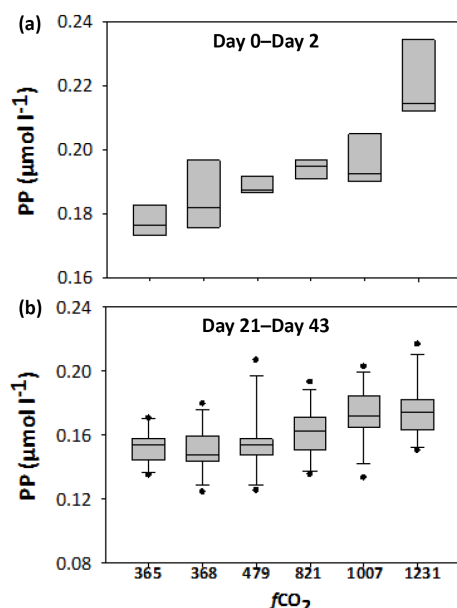


**Figure 6.** Contribution of the individual P-fractions to TP in fjord water and in the respective mesocosms. The data are averaged for the period when TP measurements were done (day –3 to day 29).

(Fig. 6). With a few exceptions,  $\text{PO}_4$  concentrations declined from the beginning of the study period until the end of phase I and increased during phase II and the beginning of phase III. These changes were caused by upwelling of  $\text{PO}_4$  enriched deep water of higher salinity and lower temperatures. The subsequent decline in  $\text{PO}_4$  between days 33 and 40 was caused by the stimulation of phytoplankton production, as indicated by the increase in Chl *a* concentration (Fig. 4).

#### 4 Discussion

An increase in  $\text{CO}_2$  in marine waters and the associated acidification may potentially have multiple effects on organisms and biogeochemical element cycling (Gattuso and Hansson, 2011). Reported findings indicate wide-ranging responses, probably depending on the investigated species and growth conditions. For example,  $\text{CO}_2$  stimulation as well as lack of stimulation were found for primary production and carbon fixation (Beardall et al., 2009; Boettjer et al., 2014), DOC release (Engel et al., 2014; MacGilchrist et al., 2014) and phytoplankton growth (Riebesell and Tortell, 2011). An interaction of  $\text{CO}_2$  effects with phosphorus and iron avail-



**Figure 7.** PP concentration in the mesocosms during the initial phase from day 0 to day 2 (a) and from day 23 until the end (b) of experiment.

ability has been found by Sun et al. (2011) for a diatom *Pseudo-nitzschia multiseriata* and by Yoshimura et al. (2014) for a diatom-dominated subarctic plankton community. Thus, responses of organisms and ecosystems to enhanced CO<sub>2</sub> concentrations are complex and still poorly understood. The present study is the first to determine the effects of increased CO<sub>2</sub> levels on phosphorus cycling in a brackish water ecosystem.

#### 4.1 Response of P-pools and P-uptake to enhanced CO<sub>2</sub> in the mesocosms

The Finnish coast off the Gulf of Finland is one of the most important upwelling regions in the Baltic Sea. During our investigation in 2012, surface temperatures, obtained from the NOAA satellite (Siegel and Gerth, 2013), showed that upwelling persisted during the whole study period but with varying intensity. The intensity of upwelling shaped the pattern of temperature not only in the fjord but also in the mesocosms varying from 7.8 to 15.9 °C. Such variations in temperature influence the phosphorus transformation and interleave with CO<sub>2</sub> effects.

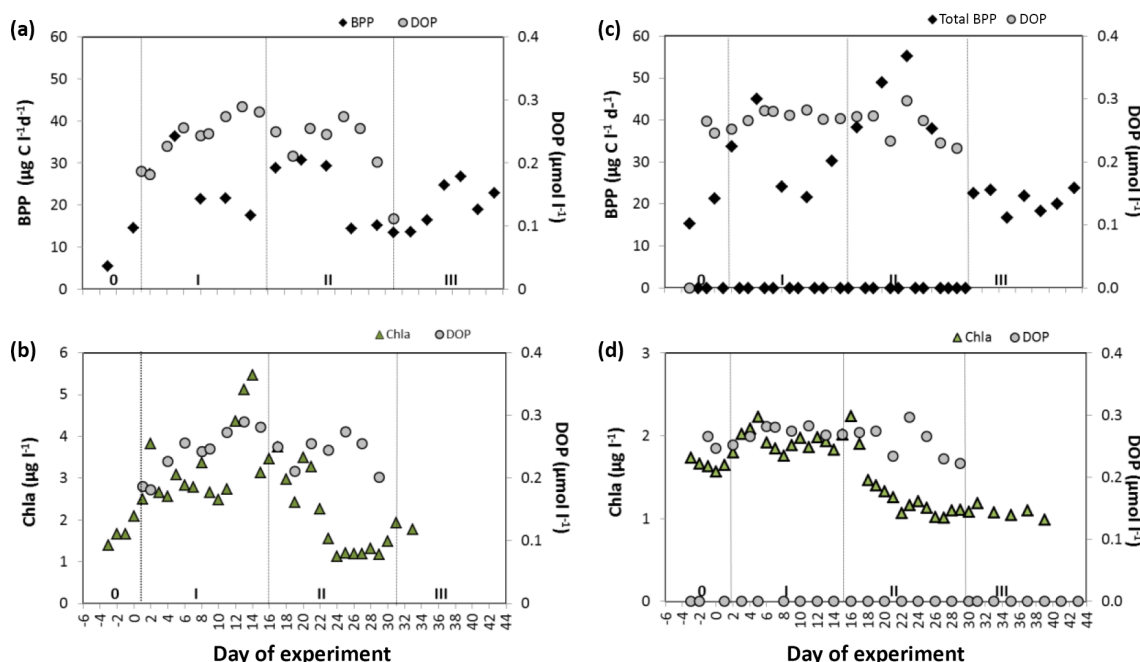
While nutrients were added in previous CO<sub>2</sub> enrichment experiments (Riebesell et al., 2008, 2013; Schulz et al., 2008), no amendments were undertaken in this study in order to be close to natural conditions. Initial PO<sub>4</sub> concentrations of only  $0.17 \pm 0.01 \mu\text{mol L}^{-1}$  were measured, however, PO<sub>4</sub> was never exhausted (Figs. 5, 6). Cellular C : P and N : P ratios were close to the Redfield ratio. Therefore, phosphorus limitation unlikely occurred in this experiment. Simultane-

ous low nitrate and ammonium concentrations (Paul et al., 2015b) formed nutrient conditions that benefit the growth of diazotrophic cyanobacteria. However, a cyanobacteria bloom failed to appear, despite the low-level presence of *Aphanizomenon* sp. and *Dolichospermum* sp. (Paul et al., 2015a) as potential seed stock. For Baltic Sea summer conditions, the phytoplankton development with maximum Chl *a* concentrations of  $2.2\text{--}2.5 \mu\text{g L}^{-1}$  remained relatively low with the highest contribution of cryptophytes and chlorophytes in phase I and at the beginning of phase II. Picoplankton was mostly the dominating size fraction, amounting to ~20–70 % of Chl *a* in phase I and up to ~85 % in phase III (Paul et al., 2015b). However, a positive correlation of *f*CO<sub>2</sub> with Chl *a* was observed only for the size fraction >20 μm. The abundance of diatoms that could be a part of this fraction increased from ~day 23 to day 30 and might have an influence on this relationship.

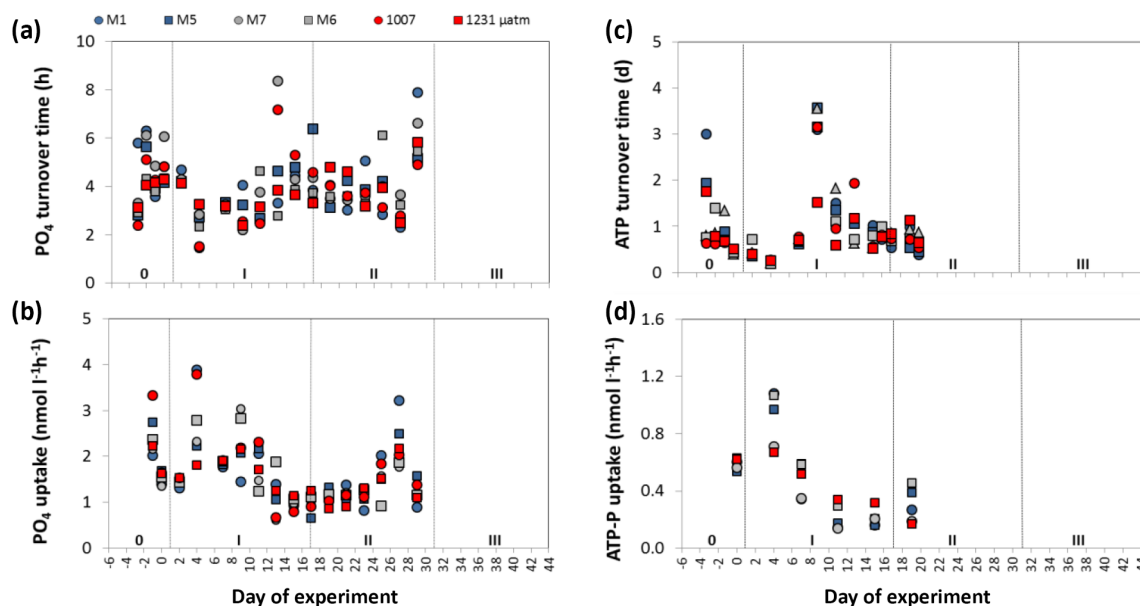
Against this background, the CO<sub>2</sub> perturbation did not cause significant changes in phosphorus pool sizes, DOP composition, and P-uptake rates from PO<sub>4</sub> and ATP when the whole study period was considered. However, small yet significant short-term effects on PO<sub>4</sub> and PP pool sizes were observed in phases I, III and partially in phase II (Fig. 7). CO<sub>2</sub> elevation stimulated the formation of PP until day 3 (Fig. 5b) when chlorophytes, cyanobacteria, prasinophytes and the pico-cyanobacteria started to grow (Paul et al., 2015b).

The effects of CO<sub>2</sub> addition on PO<sub>4</sub> and PP pool sizes were evident from day 23 onwards (Figs. 5b, 7). PO<sub>4</sub> concentrations were slightly, but significantly lower in the high CO<sub>2</sub> treatment than in the untreated mesocosms, accompanied by significantly elevated PP concentrations. This indicates that the transformation of PO<sub>4</sub> into PP was likely stimulated under high CO<sub>2</sub> conditions. Since Chl *a* was also elevated at similar PP : Chl *a* ratios, the PO<sub>4</sub> taken up was used for new biomass formation. However, the elevated transformation of PO<sub>4</sub> into PP was not reflected in the PO<sub>4</sub> uptake rates which can be seen as gross uptake rates. But, an increase of PP, caused by biomass formation, while the PO<sub>4</sub> uptake remained unchanged can only occur when the P release from organisms is reduced. Thus, it is likely the net uptake was modified under CO<sub>2</sub> elevation and not the gross uptake.

While in phases II and III, high CO<sub>2</sub> levels caused a change in the PP and PO<sub>4</sub> pools for about 22 days, changes lasting only 2 days have been observed at the beginning of phase I (Fig. 7a), but, shorter effects cannot be excluded. Uptake and release are assumed to be continuous processes and can alter the P pool sizes on timescales shorter than 1 day. Thus, variations and differences in the treatments can be overseen at daily sampling. Unger et al. (2013) demonstrated that an accelerated PO<sub>4</sub> uptake by the cyanobacterium *Nodularia spumigena* under elevated CO<sub>2</sub> incubations could only be observed during the first hours. Thereafter, the differences were balanced and the same level of radiotracer labelling was reached in all treatments. An acceleration in the formation of particulate P under CO<sub>2</sub> elevation without any



**Figure 8.** Development of DOP in relation to bacterial production (BPP) and phytoplankton biomass (Chl *a*) in the fjord (a, b) and in the mesocosms (c, d). For mesocosms, mean values averaged over all treatments are given.

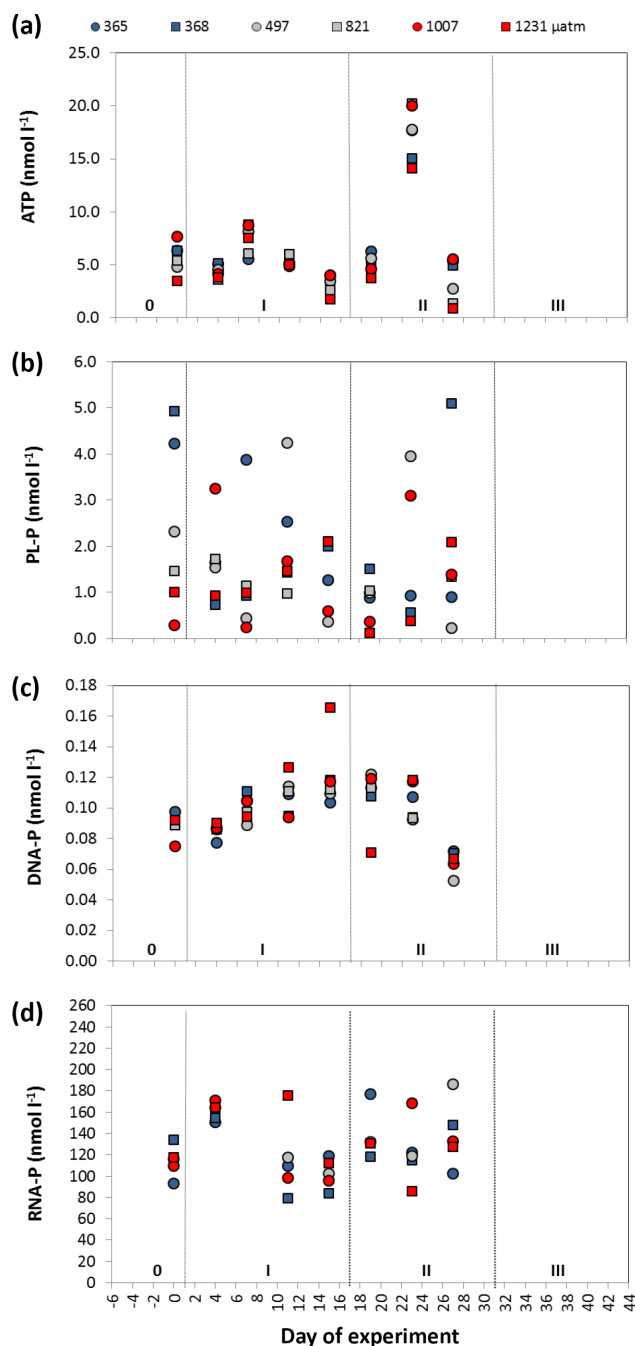


**Figure 9.** Turnover times of PO<sub>4</sub> (a) and ATP (c) in the mesocosms as well as the respective uptake rates (b, d).

changes of PO<sub>4</sub> turnover times was also observed by Tanaka et al. (2008). They observed an increase of the PP amount and an earlier appearance of the PP maximum under CO<sub>2</sub> elevation.

Correlations calculated by using the Spearman rank test between P pools or uptake rates and other parameters for each mesocosm are presented in Table 2. The relationships

between PP and TP with Chl *a* disappeared at elevated *f*CO<sub>2</sub>, whereas correlations developed between PP and PC as well as between the PO<sub>4</sub> uptake by phytoplankton in the > 3  $\mu\text{m}$  size class and the PP:Chl *a* ratio (Table 2). These shifts could be caused by changes in the phytoplankton composition deduced from CO<sub>2</sub> effects on the pigment composition (Paul et al., 2015b).



**Figure 10.** Development of DOP compounds in the mesocosms and in the fjord from day 0 to day 27.

Independent of the CO<sub>2</sub> treatment, TP decreased by  $2.6 \text{ nmol L}^{-1} \text{ day}^{-1}$  in all mesocosms over the course of the experiment, in agreement with the measured sedimentation rates (Paul et al., 2015b). The strongest decrease ( $\sim 3.2 \text{ nmol L}^{-1} \text{ day}^{-1}$ ) occurred during phase I. Of the total TP removal during this phase ( $48 \text{ nmol L}^{-1}$ ), 84 % ( $\sim 40.5 \text{ nmol L}^{-1}$ ) could be explained by the decrease in PP and 16 % ( $\sim 8 \text{ nmol L}^{-1}$ ) by changes in the dissolved pool.

However, the PO<sub>4</sub> decline ( $\sim 34.5 \text{ nmol L}^{-1}$ ) was stronger than that of the total dissolved P pool since DOP increased in parallel by  $\sim 26.5 \text{ nmol L}^{-1}$ . Thus, about 77 % of the PO<sub>4</sub> reduction was retrieved as DOP and remained in the dissolved P-pool as the main pathway of PO<sub>4</sub> transformation.

## 4.2 Phosphorus dynamics in the Storfjärden

Nutrients in upwelled water during our study were depleted in dissolved inorganic nitrogen and enriched in PO<sub>4</sub>, as reported for other upwelling areas of the Baltic Sea (Lass et al., 2010). Thus, ammonium and NO<sub>2</sub>+NO<sub>3</sub> concentrations in the surface water were only in the nanomolar range (Paul et al., 2015b). PO<sub>4</sub> increased in parallel with the increase in salinity and decrease in temperature. Maximum PO<sub>4</sub> concentrations of  $0.33$  and  $0.42 \mu\text{mol L}^{-1}$  (Figs. 5, 6) were observed at the end of the upwelling events in phases 0 and II, respectively. The correlation with Chl *a* and PP indicated that PO<sub>4</sub> was utilized during plankton growth in the subsequent relaxation phases I and III. Due to PO<sub>4</sub> input into surface water, the phytoplankton community was unlikely P-limited indicated by PC : PP ratios of 86–189 (mean 125,  $n = 23$ ; Paul et al., 2015b). However, the PO<sub>4</sub> availability might be not the only reason for the good P-nutritional status of the plankton. It can be deduced from the long PO<sub>4</sub> turnover times in the mesocosms, where external input was excluded, that the P demand of the plankton community might be low. The P content deduced from PP : Chl *a* ratios of  $0.05$ – $0.15 \mu\text{mol P} (\mu\text{g Chl } a)^{-1}$  was somewhat lower than those observed during an upwelling event along the east coast of Gotland, where ratios between  $0.1$  and  $0.2 \mu\text{mol P} (\mu\text{g Chl } a)^{-1}$  (Nausch et al., 2009) were estimated.

PP concentrations of  $0.13$ – $0.3 \mu\text{mol L}^{-1}$  were in the range typically observed in the Baltic Proper (Nausch et al., 2009, 2012). However, PP concentrations in the Gulf of Finland may reach higher values, as was the case in the summer of 2008, when the observed PP concentration was  $0.35 \pm 0.07 \mu\text{mol L}^{-1}$  (Nausch and Nausch, 2011).

DOP concentration of  $0.27 \pm 0.02 \mu\text{mol L}^{-1}$  during our study was similar to that detected in the Gulf of Finland in the summer of 2008 (Nausch and Nausch, 2011). In the Baltic Sea, DOP exhibits vertical gradients with maximum concentrations in the euphotic surface layer and lower than  $0.1 \mu\text{mol L}^{-1}$  at depths below 25 m. Thus, the observed DOP dynamics in surface water during our study can be assumed to be the result of release, consumption and mineralization by organisms or input from land. The relationship of DOP with Chl *a* and BPP (Fig. 8) indicated that the increased DOP concentrations in phase I may be due to release by phytoplankton supplemented by bacterial release. DOP can be accumulated in water only when the release exceeded the consumption or degradation. During phase II, phytoplankton biomass was low and DOP release should thus be minor. Since the small mesozooplankton increased in the fjord similar to those reported for the mesocosms in phases II and III (Paul et al.,

2015b) DOP could be released during grazing combined with the observed temporal offset of BPP and DOP maxima. Thus, the observed DOP variations may be the result of processes in the surface water.

## 5 Conclusions

Surface water in Storfjärden showed highly variable  $f\text{CO}_2$  conditions and reached levels of up to 800  $\mu\text{atm}$ , which is similar to that expected in ca. 100 years from now. Deduced from the high frequency of upwelling events, organisms experience elevated  $f\text{CO}_2$  more or less regularly. Thus, a general impact of  $f\text{CO}_2$  on P pools and P uptake rates in the mesocosms could not be identified for the overall period of investigation. However, short-term responses to  $f\text{CO}_2$  elevation lasting only few days were observed for the transformation of  $\text{PO}_4$  into PP that was linked with stimulation of phytoplankton growth. Although statistically significant, it is difficult to assess if the differences between the treatments are of ecological relevance. Such short-term variations are possible in the phosphorus dynamics since the pools size can be transformed within hours and there changes are in the nanomolar concentration range. There are also indications that relationships of P pool sizes or uptake with Chl *a* and PC can change as  $f\text{CO}_2$  increases, but the underlying mechanisms are still unclear. The transformation of  $\text{PO}_4$  into DOP was not affected by CO<sub>2</sub> elevation. It may be the major pathway of phosphorus cycling under hydrographical and phytoplankton growth conditions as occurred in our experiment.

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